

118. The Synthesis of (\pm)-11-Deoxydaunomycinone via Regioselective Tandem *Diels-Alder* Reactions¹⁾

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The 2,5-dimethylidene-3,6-bis[(*Z*)-(2-nitrophenyl)sulfonylmethylidene]-7-oxabicyclo[2.2.1]heptane (**13**) can be used to generate polyfunctional and multicyclic molecules with high regio- and stereoselectivity *via* two successive *Diels-Alder* additions using two different dienophiles. This principle has been applied to the synthesis of (\pm)-11-deoxydaunomycinone (**7**), the aglycone of an important antitumor drug. The 2,3-didehydroanisole adds to **13** and gives the monoadduct **14** with high regioselectivity. No trace of bis-adduct is observed. The 1,4-epoxy-1,2,3,4-tetrahydro-5-methoxy-3-methylidene-2-[(*Z*)-(2-nitrophenyl)sulfonylmethylidene]anthracene (**15**) obtained on treating **14** with K_2CO_3 adds to methyl vinyl ketone to give [(1*RS*,2*SR*,5*RS*,12*RS*)-5,12-epoxy-1,2,3,4,5,12-hexahydro-7-methoxy-1-(2-nitrophenyl)sulfonyl-2-naphthacenyl] methyl ketone (**16**) with high regio- and stereoselectivity. The acid-catalyzed 7-oxanorbornadiene \rightarrow phenol rearrangement of **16** is regioselective and gives (5-acetoxy-3,4-dihydro-7-methoxy-2-naphthacenyl) methyl ketone (**20**) which was transformed into (\pm)-7,11-dideoxydaunomycinone ((\pm)-**24**), a known precursor of **7**.

Introduction. – The anthracyclines constitute a group of natural antibiotics isolated from cultures of various *Streptomyces* [1] spp., certain members of which possess significant antineoplastic activity. Among these, daunomycin (**1**) and adriamycin (**2**) have been the object of considerable interest in recent years because of their value in cancer chemotherapy [2] which was only limited because of their cardiotoxicity [3]. Total syntheses of the corresponding aglycone parts of daunomycinone (**3**) and adriamycinone (**4**) have been subject of intense study in the last 15 years due to the lack of an efficient biosynthetic process [4] as well as a search for more active analogs with reduced cardiotoxicity [5]. *Arcamone* and coworkers [6] isolated from *Micromonospora peucetica* the 11-deoxydaunomycin (**5**) and 11-deoxyadriamycin (**6**) which were shown to possess significant antitumor activity and to be less cardiotoxic than **1** and **2**. The potential advantages associated with these compounds have prompted interest in their preparations, and several total syntheses of the aglycone fragment (\pm)-11-deoxydaunomycinone

1^a) $R^1 = H, R^2 = OMe, R^3 = OH, X = H, Z = \text{'daunosaminyl'}$; daunomycin

2^a) $R^1 = H, R^2 = OMe, R^3 = OH, X = OH, Z = \text{'daunosaminyl'}$; adriamycin

3 $R^1 = H, R^2 = OMe, R^3 = OH, X = Z = H$; daunomycinone

4 $R^1 = H, R^2 = OMe, R^3 = OH, X = OH, Z = H$; adriamycinone

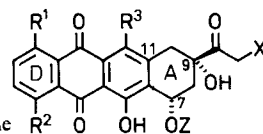
5 $R^1 = H, R^2 = OMe, R^3 = X = H, Z = \text{'daunosaminyl'}$; 11-deoxydaunomycin

6 $R^1 = H, R^2 = OMe, R^3 = H, X = OH, Z = \text{'daunosaminyl'}$; 11-deoxyadriamycin

7 $R^1 = H, R^2 = OMe, R^3 = H, X = Z = H$; 11-deoxydaunomycinone

8 $R^1 = R^2 = H, R^3 = OH, X = Z = H$

9 $R^1 = R^2 = OMe, R^3 = OH, X = Z = H$

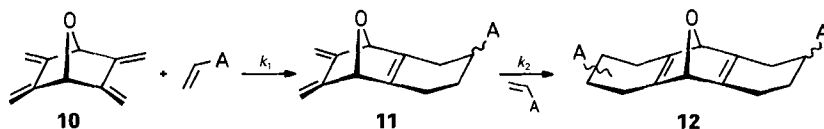


^a) Customary numbering of this class of compounds.

¹⁾ Part of the Ph. D. thesis of Jean-Marc Tornare, University of Lausanne, March 1985.

(7) have been published [7–20]. *Kishi* and coworkers [21] have proposed a practical asymmetric synthesis of 7 and related compounds [22].

A few years ago we reported that 2,3,5,6-tetrakis(methylidene)-7-oxabicyclo[2.2.1]heptane (**10**), readily obtained from furan and maleic anhydride [23], can be used to prepare various anthracyclinones such as **3** [24], **8** [25], and **9** [26]. The principle of our strategy rests upon the fact that the rate constant of the *Diels-Alder* addition of **10**→**11** (k_1) is much larger than that (k_2) for the reaction of the corresponding monoadduct **11** with the same dienophile giving the corresponding bis-adduct **12** [27]. A major difficulty in the synthesis of the anthracyclinones **3**, **4**, and **7** is to control the substitution pattern of the two remote rings A and D. This cannot be realized following our strategy (see, however, [24]) **10**→**11**→**12**, unless one utilizes an analog of **10** with substituted diene

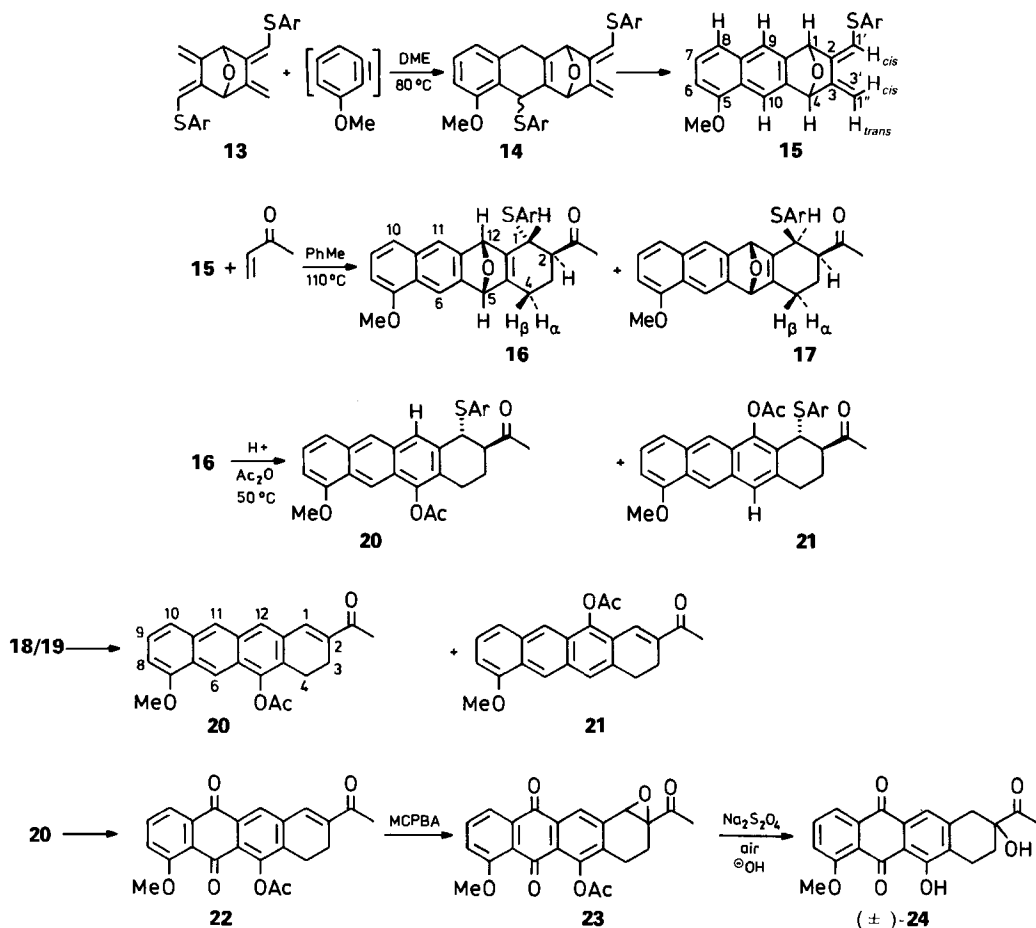


moieties whose substituents are able to steer the regiochemistry of the two successive (tandem) cycloadditions²). Recently we have succeeded in preparing the tetraene **13** (Ar = 2-NO₂C₆H₄) [29]. Double addition of 2-nitrobenzenesulfonyl chloride to **10**, followed by elimination of two equiv. of HCl afforded **13** in a stereoselective fashion. The tetraene **13** was shown to be a versatile reagent for regio- and stereoselective tandem *Diels-Alder* reactions [29]. As for **10**, the rate of the addition of the first equiv. of a strong dienophile to **13** is much larger than the rate of the addition of the second equiv. of the same dienophile to the corresponding monoadduct. We report here on the application of tetraene **13** to the doubly-convergent regioselective synthesis of (±)-11-deoxydaunomycinone (**7**).

Results and Discussion. – The deamination of 2-amino-6-methoxybenzoic acid [30] (pentyl nitrite, dimethoxyethane (DME), 85 °C [31]) in the presence of tetraene **13** gave a mixture of the adduct **14** and its product of elimination **15**. On treating this mixture with K₂CO₃ in boiling acetone, **15** was the sole product isolated in 46% yield. The reaction of the 2,3-didehydroanisole (assumed to be generated during the deamination reaction) with **13** is highly regioselective since no trace of an other regioisomer than **15** could be observed. Furthermore, the reaction of **14** or/and **15** with 2,3-didehydroanisole is much slower than with **13** as no trace of the corresponding bis-adduct could be detected. This is in contrast with the reaction of didehydrobenzene with tetraene **10** which was shown to give the corresponding mono- and bis-adducts competitively [25]. The modest yield of the transformation **13**→**15** is attributed to a certain lability of the tetraene **13** and diene **15** under the reaction conditions.

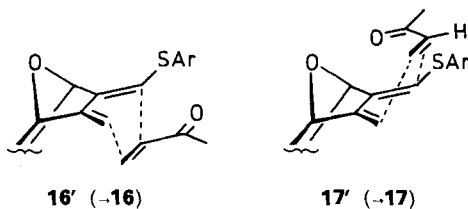
The structure of **15** was established by 360-MHz-¹H-NMR spectroscopy and by measurement of NOE's between adjacent protons. Irradiation of the olefinic proton at 6.43 ppm (H_{cis}-C(1'); *cis* with respect to C(2), C(3)) led to a NOE at 5.43 ppm attributed to H_{cis}-C(1''). No NOE was observed at H-C(1) (6.23 ppm), in agreement

²) Tetraenes substituted at a bridgehead centre such as 1-(dimethoxy)methyl-2,3,5,6-tetramethylidene-7-oxabicyclo[2.2.1]heptane can also be considered as reagents for tandem regio- and stereo-selective tandem cycloadditions, see [28].



with the (*Z*)-configuration of the thioolefin moiety. Irradiation of the signal H_{cis}-C(1'') led to NOE's at 6.43 (H_{cis}-C(1')) and 5.35 ppm (H_{trans}-C(1'')). Irradiation of the latter signal gave NOE's at 5.43 (H_{cis}-C(1'')) and 5.82 ppm (H-C(4)). Irradiations of the signals at 5.82 (H-C(4)), 8.21 (H-C(10)), and 6.23 (H-C(1)) ppm led to NOE's at 8.21 (*s*, H-C(10)), 5.82 (*br. s*, H-C(4)), and 7.73 ppm (*s*, H-C(9)), respectively. Irradiation at 7.73 ppm (H-C(9)) led to NOE's at 6.23 (H-C(1)) and 7.40 ppm (H-C(8)). The latter signal appeared as a *dd* with ³*J* = 7.5 and ⁴*J* = 2.8 Hz.

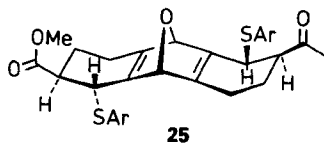
The addition of methyl vinyl ketone (MVK) to **15** in toluene (110°C) gave a 9:1 mixture of the adducts **16** and **17** (88% isolated). The major isomer **16** was isolated in pure form by recrystallization from CH₂Cl₂/Et₂O 4:1. No trace of any other isomeric adduct could be detected, thus demonstrating the completely regioselective nature and the high stereoselectivity of the *Diels-Alder* reactions **15** + MVK. Out of eight possible orientations for the cycloadducts (regio-isomers '*ortho*' vs. '*meta*', face selectivity '*exo*' vs. '*endo*', *Alder*- vs. *anti-Alder*-rule orientation of the dienophile), only those represented by the transition states **16'** and **17'** are favoured. In contrast with the '*exo*'-face selectivity of the cycloadditions of ethylenetetracarboxytrile and *N*-phenyltriazolinedione to 2,3-dimethylidene-7-oxabicyclo[2.2.1]heptanes substituted with a chlorine atom or a methoxy



group at the diene moiety [32], the thermal reaction **15** + MVK is 90% 'endo'-face selective. Interestingly, the 'endo'-face attack (giving **16**) follows the *anti-Alder*-rule of cycloaddition, whereas the 'exo'-face attack (giving **17**) obeys the *Alder* rule [33]. Competition between a steric factor (favouring 'exo'-face attack) and a stereoelectronic factor (favouring 'endo'-face attack) can be invoked to interpret our observations³⁾.

The structures of adducts **16** and **17** were deduced from their 360-MHz ¹H-NMR spectra and from double-irradiation experiments.

The α -position of the arylthio substituent at C(1) was given by the homoallylic coupling constants [32] [36] measured between the bridgehead proton H-C(5) and H-C(1) ($^2J = 1.0$ Hz in **16**; < 0.2 Hz in **17**⁴⁾) and between H $_{\beta}$ -C(1)/H $_{\beta}$ -C(4) ($^5J = 2.2$ Hz in **16**) and H $_{\beta}$ -C(1)/H $_{\alpha}$ -C(4) ($^5J = 2.8$ Hz in **16**). The proton H $_{\beta}$ -C(4) of **16** was recognized by its long-range coupling with the bridgehead proton H-C(12) ($^5J = 1.0$ Hz) whereas H $_{\alpha}$ -C(4) does not couple with H-C(12). The β -position of the acetyl side chain at C(2) was deduced from the vicinal coupling constants measured between H-C(2) and H-C(1) (6.2 Hz in **16**, 4.1 Hz in **17**⁴⁾), between H-C(2) and H $_{\alpha}$ -C(3) (4.0 Hz in **16**), between H-C(2) and H $_{\beta}$ -C(3) (7.8 Hz in **16**). The structure of **16** (and **17**) was further confirmed by comparing the ¹H-NMR characteristics with those observed for the bis-adduct **25** whose structure was established by X-ray crystallography [34].



On treating **16** in Ac₂O with a small amount of CF₃COOH (80°C, 1 h), a 3:1 to 4:1 mixture of the tetrahydronaphthacene derivatives **18** and **19** was obtained⁵⁾. The crude mixture was then treated with *meta*-chloroperbenzoic acid (MCPBA; NaHCO₃, CH₂Cl₂, 20°C, 15 min) to afford a mixture **20/21** from which the major isomer **20** was obtained pure (65% based on **16**) by crystallization from CH₂Cl₂/Et₂O 5:1. The structure of **20** was given by its mode of formation and its 360-MHz ¹H-NMR spectra. It was confirmed by NOE measurements between the adjacent protons (see *Exper. Part*). Jones' oxidation (4N CrO₃ in aqueous H₂SO₄) of **20** in acetone (0°C, N₂ atmosphere, 1 h) gave the anthraquinone **22** in 64% yield. The latter was epoxidized (MCPBA, 10 molar excess, CH₂Cl₂, 15 h) to **23** (42%) at 20°C. At higher temperature, *Baeyer-Villiger*-reaction products were formed competitively. Attempts to epoxidize **22** with basic H₂O₂ led to decomposition of the product. The reduction of the epoxide **23** with Na₂S₂O₄ under alkaline conditions [37] yielded the known (\pm)-7,11-dideoxydaunomycinone ((\pm)-**24**) [7e] [10] [12c] which was identical (spectral data, mixed 360-MHz ¹H-NMR spectrum, mixed m.p.) with an authen-

³⁾ Further examples of face-selective cycloadditions of **13** and related 2,3-dimethylidenebicyclo[2.2.1]heptanes will be presented in forthcoming publications [34] [35].

⁴⁾ Measured in the ¹H-NMR spectrum of a mixture **16/17** (mother liquor of the crystallization of **16**).

⁵⁾ In Ac₂O/CH₂Cl₂ 1:1, a 1:1 mixture **18/19** was obtained.

tic sample of (\pm)-**24** furnished by Dr. Gesson⁶). The conversion of (\pm)-**24** into (\pm)-11-deoxydaunomycinone ((\pm)-**7**) has been reported [7] [11].

Conclusion. – The arylthio substituents in the tetraene **13** add further flexibility to our doubly-convergent approach⁷ to the synthesis of anthracyclonones. They render the tandem *Diels-Alder* additions completely regioselective, even with a highly reactive dienophile such as 2,3-didehydroanisole [39]. The arylthio substituents are also useful in the further transformations of the adducts and bis-adducts. For instance, base-induced elimination of arenesulfenic acid in **14** allows aromatization of the C-ring of the anthracyclonone. In the bis-adduct **16**, the arylthio substituent makes the acid-catalyzed oxanorbornadiene \rightarrow phenol rearrangement [40] regioselective. In principle, and following techniques applied by us earlier [24–26], compounds **18** and **19** or/and **20** and **21** could be used to generate daunomycinone (**3**). Work is underway to prepare **3** and **7** in an optically pure form⁸).

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Experimental Part

General Remarks. See [25] [42]. None of the procedures described below has been optimized.

2,5-Dimethylidene-3,6-bis[(*Z*)-(2-nitrophenyl)sulfonylmethylidene]-7-oxabicyclo[2.2.1]heptane (**13**). LiCl (17.4 g, 0.41 mol) and 2-nitrobenzenesulfonyl chloride (17.1 g, 0.09 mol) were dissolved in anh. AcOH (200 ml) at 20°. A soln. of 2,3,5,6-tetramethylidene-7-oxabicyclo[2.2.1]heptane (5.8 g, 0.0397 mol) [23] in anh. AcOH (90 ml) was added and the mixture stirred at 20° for 4 h. The precipitate *A* (8.3 g of unstable bis-adducts, 40%) was filtered off and used directly for the elimination of HCl. The filtrate was stirred at 20° for 12 h and yielded a second precipitate *B* of bis-adducts (7 g, 34%). The precipitates were dried separately *in vacuo* at 20°. *A* and *B* were dissolved in 265 and 225 ml of anh. THF, respectively. After cooling to –78°, *t*-BuOK (3 mol-equiv.) was added portionwise under stirring and Ar. The temp. was allowed to raise slowly to 20°. After addition of H₂O (100 ml), the mixtures were extracted with CH₂Cl₂ (200 ml, 3 times). The extracts were washed with H₂O (200 ml, 3 times) and filtered through silica gel (100 g). After solvent evaporation, the residues were recrystallized from CH₂Cl₂/hexane 9:1. *A* yielded pure **13** (3.9 g, 24% based on **10**), yellow crystals, m.p. 187–188°. UV (CH₃CN): 240 (34 000), 300 (20 500), 382 (12 500). IR (KBr): 3100, 3000, 1595, 1570, 1505, 1455, 1345, 1305, 1255, 1225, 1170, 1150, 1110, 1060, 1045, 975, 915. ¹H-NMR (CDCl₃): 8.25–7.34 (*m*, 8 arom. H); 6.49 (*s*, 2 ArSCH); 5.67 (*br. s*, H–C(1), H–C(4)); 5.45 (*s*, HCH=C(2) and HCH=C(5) *cis* to C(2), C(3) and C(5), C(6), resp.); 5.37 (*s*, 2H). ¹³C-NMR (CDCl₃): 146.6 (*s*, C(3), C(6)); 146.0 (*s*, C(arom.)–NO₂), 143.9 (*s*, C(2), C(5)); 135.7 (*s*, C(arom.)–S); 133.7, 128.9, 125.9 (2 C) (*dd*, ¹J_{C,H} = 170, ³J_{C,H} = 8, 4 arom.); 110.5 (*d*, ¹J_{C,H} = 178, ArSCH); 104.6 (*t*, ¹J_{C,H} = 164, H₂C=C(3), H₂C=C(6)); 83.6 (*dm*, ¹J_{C,H} = 170, C(1), C(4))⁹. MS (70 eV): 453 (11), 452 (34, *M*⁺), 336 (12), 333 (27), 314 (30), 298 (31), 253 (16), 252 (24), 236 (15), 225 (13), 224 (29), 223 (19), 222 (16), 221 (16), 208 (14), 195 (14), 179 (23), 154 (27), 147 (37), 138 (76), 44 (100). Anal. calc. for C₂₂H₁₆N₂O₃S₂ (452.512): C 58.39, H 3.56; found: C 58.21, H 3.57.

Precipitate *B* yielded a mixture of **13** (35–45%), 2,5-dimethylidene-3-[(*Z*)-(2-nitrophenyl)sulfonylmethylidene]-6-[(*E*)-(2-nitrophenyl)sulfonylmethylidene]-7-oxabicyclo[2.2.1]heptane (**26**; 5–15%) and 2-endo-chloromethyl-3,6-dimethylidene-2-exo-(2-nitrophenyl)sulfonyl-5-[(*Z*)-(2-nitrophenyl)sulfonylmethylidene]-7-oxabicyclo[2.2.1]heptane (**27**; 5–15%) separated by column chromatography on silica gel (500 g, AcOEt/hexane

⁶) We thank Dr. *J.P. Gesson* for ¹H-NMR and IR spectra and an authentic sample of (\pm)-**24**.

⁷) It has been called a 'doubly, doubly-convergent' synthesis [38].

⁸) See *e.g.*, the techniques applied to the synthesis of (+)-4-demethoxydaunomycinone [37] [41].

⁹) The signal attributions were confirmed by 2-D (δ_C vs. δ_H) NMR spectroscopy. We thank Mr. *J. Wernly* for these measurements.

1:3). Characteristics of **26**: unstable, yellow oil. $^1\text{H-NMR}$ (CDCl_3): 8.26–7.32 (*m*, 8 arom. H); 6.55, 6.48 (2s, 2 ArSCH); 6.0, 5.61, 5.44, 5.24 (4s, 4 olef. H); 5.64, 5.22 (2 br. s, H–C(1), H–C(4)); NOE measurements confirmed the attribution and the structure.

Characteristics of **27**: yellow crystals, m.p. 192–193° (CH_2Cl_2 /hexane 9:1). UV (CH_3CN): 240 (sh, 20000), 300 (12500), 376 (6000). IR (KBr): 3100, 3080, 3020, 2960, 1590, 1570, 1525, 1455, 1430, 1365, 1340, 1305, 1290, 1260, 1210, 1110, 1060, 985, 960, 915, 900. $^1\text{H-NMR}$ (CDCl_3): 8.25–7.35 (*m*, 8H); 6.65 (*s*, 1H); 5.42 (br. s, 1H); 5.35 (*s*, 1H); 5.22 (*s*, 1H); 5.13 (*s*, 1H); 5.05 (br. s, 1H); 5.02 (br. s, 1H); 3.82, 3.69 (2*d*, $^2J = 12.5$, ClCH₂). $^{13}\text{C-NMR}$ (CD_2COCD_2): 156.4, 150.6, 147.4, 146.4, 144.0, 135.8 (br. s); 139.7, 135.1, 132.7, 131.4, 130.3, 127.5, 126.6, 124.8 (*dd*, $^1J_{\text{C,H}} = 170$, $^3J_{\text{C,H}} = 8$); 126.3 (*s*); 115.7 (*d*, $^1J_{\text{C,H}} = 180$); 110.9, 104.1 (2*t*, $^1J_{\text{C,H}} = 164$); 87.5, 86.7 (2 *dm*, $^1J_{\text{C,H}} = 172$); 65.2 (*s*); 50.1 (*t*, $^1J_{\text{C,H}} = 156$). MS (70 eV): 336 (7.2, M^+ – [$\text{SC}_6\text{H}_4\text{NO}_2$]), 334 (19), 298 (4), 252 (10), 236 (7), 225 (6), 224 (10), 223 (11), 208 (8), 197 (28), 196 (11), 195 (73), 179 (18), 160 (9), 147 (24), 138 (100). Anal. calc. for $\text{C}_{22}\text{H}_{17}\text{N}_2\text{O}_5\text{S}_2$ (488.970): C 54.04, H 3.50; found: C 54.25, H 3.60.

The configuration of C(2) in **27** is not established unambiguously.

When less than 2.2 mol-equiv. of 2-nitrobenzenesulfonyl chloride was used in the preparation of **13**, the treatment of precipitate **B** with *t*-BuOK gave a mixture of **13**, **26**, **27**, and 2,3,5-trimethylidene-6-[(*Z*)-(2-nitrophenyl)sulfonylmethylidene]-7-oxabicyclo[2.2.1]heptane (**28**), separated by column chromatography on silica gel (AcOEt/hexane 1:3). Characteristics of **28**: yellow crystals, m.p. 138–139° (CH_2Cl_2 /hexane 9:1). UV (CH_3CN): 230 (25000), 246 (19000), 308 (9500), 374 (5500). IR (KBr): 3100, 3000, 2820, 1590, 1570, 1510, 1450, 1335, 1305, 1255, 1150, 1105, 1040, 900, 850. $^1\text{H-NMR}$ (CDCl_3): 8.23–7.31 (*m*, 4 arom. H); 6.44 (*s*, ArSCH); 5.58 (br. s, H–C(1)); 5.41 (*s*, 1H); 5.35 (*s*, 2H); 5.24 (*s*, 2H); 5.19 (*s*, 1H); 5.16 (br. s, H–C(4)). $^{13}\text{C-NMR}$ (CDCl_3): 147.2, 145.8, 145.6, 145.4, 144.0, 135.9 (br. s); 133.6, 128.7, 125.8, 125.7 (*dd*, $^1J_{\text{C,H}} = 168$, $^3J_{\text{C,H}} = 8$, arom. C); 109.6 (*d*, $^1J_{\text{C,H}} = 176$); 103.6, 103.2 (*t*, $^1J_{\text{C,H}} = 162$); 85.7, 83.3 (*dm*, $^1J_{\text{C,H}} = 166$, C(1), C(4)). CI-MS (CH_3): 300 (100, M^+ + 1), 258 (19), 156 (13), 154 (25). Anal. calc. for $\text{C}_{16}\text{H}_{16}\text{NO}_3\text{S}$ (299.346): C 64.19, H 4.37; found: C 64.09, H 4.27.

1,4-Epoxy-1,2,3,4-tetrahydro-5-methoxy-3-methylidene-2-[(*Z*)-(2-nitrophenyl)sulfonylmethylidene]anthracene (**15**). A soln. of 1-amino-6-methoxybenzoic acid [30] (449 mg, 2.68 mmol) in DME (5.5 ml) and a soln. of pentyl nitrite (0.472 ml, 3.54 mmol) in DME (5.5 ml) were added dropwise and simultaneously to a stirred soln. of **13** (734 mg, 1.62 mmol) in DME (13 ml) heated to 85°. The mixture was stirred at 85° for 30 min until the end of gas evolution. After cooling to 20°, CH_2Cl_2 (80 ml) was added. The soln. was washed with 10% aq. KOH soln. (30 ml) and the aq. layer extracted with CH_2Cl_2 (30 ml, 3 times). The org. phases were united and washed with H_2O (30 ml, 5 times). After drying (MgSO_4), the soln. was filtered through silica gel (20 g), evaporated to dryness, and the residue dried *in vacuo* (P_2O_5). The oily product was dissolved in anh. acetone (30 ml). After addition of anh. K_2CO_3 (700 mg, 5 mmol), the soln. was heated under reflux for 12 h. After cooling to 20°, the mixture was filtered and the filtrate evaporated to dryness. The crude **15** was dissolved in CH_2Cl_2 (30 ml) and washed with H_2O (20 ml, 3 times). After drying (MgSO_4), the solv. was evaporated and the residue recrystallized from CH_2Cl_2 /hexane/ Et_2O 6:3:1 yielding 301 mg (46%) of yellow crystals, m.p. 186–187°. UV (CH_3CN): 216 (45000), 253 (36000), 296 (15500), 381 (4500). IR (KBr): 3000, 2960, 2920, 2840, 1730, 1710, 1595, 1570, 1520, 1505, 1450, 1435, 1335, 1300, 1280, 1260, 1255, 1215, 1190, 1180, 1130, 1100, 1075, 1060, 1030, 855. $^1\text{H-NMR}$ (CDCl_3): 8.26–7.32 (*m*, 6H, ArS, H–C(7), H–C(8)); 8.21 (*s*, H–C(10)); 7.73 (*s*, H–C(9)); 6.86 (*dd*, $^3J(\text{H–C}(6), \text{H–C}(7)) = 5.8$, $^4J(\text{H–C}(6), \text{H–C}(8)) = 2.8$, H–C(6)); 6.43 (*s*, ArSCH); 6.23 (*s*, H–C(1)); 5.82 (*s*, H–C(4)); 5.43 (*s*, HCH=C(3) *cis* to C(2), C(3)); 5.35 (*s*, HCH=C(3) *trans* to C(2), C(3)); 4.01 (*s*, CH_3O). $^{13}\text{C-NMR}$ (CDCl_3): 155.8, 147.4, 145.8, 144.7, 141.0, 140.6, 136.2, 134.2 (*s*, C(4a), C(8a), C(9a), C(2), C(3), C(5), SAR); 133.6, 128.7, 125.9, 125.6 (*dd*, $^1J_{\text{C,H}} = 166$, $^3J_{\text{C,H}} = 8$, H–C of SAR); 126.3 (*d*, $^1J_{\text{C,H}} = 160$, C(7)); 125.0 (*s*, C(10a)); 120.7 (*dm*, $^1J_{\text{C,H}} = 160$, C(8)); 118.3 (*d*, $^1J_{\text{C,H}} = 162$, C(9)); 112.8 (*d*, $^1J_{\text{C,H}} = 164$, C(10)); 109.7 (*d*, $^1J_{\text{C,H}} = 176$, ArSCH); 104.7 (*d*, $^1J_{\text{C,H}} = 158$, C(6)); 104.2 (*t*, $^1J_{\text{C,H}} = 160$, $\text{H}_2\text{C}=\text{C}(3)$); 83.9 (*d*, $^1J_{\text{C,H}} = 166$, C(1)); 81.33 (*d*, $^1J_{\text{C,H}} = 166$, C(4)); 55.5 (*q*, $^1J_{\text{C,H}} = 143$, CH_3O). MS (70 eV): 404 (11), 403 (34, M^+), 341 (10), 295 (11), 282 (11), 281 (31), 266 (20), 265 (58), 264 (21), 252 (22), 250 (39), 249 (100). Anal. calc. for $\text{C}_{23}\text{H}_{17}\text{NO}_4\text{S}$ (403.453): C 68.47, H 4.24; found: C 68.53, H 4.31.

[(1RS,2RS,5RS,12RS)-5,12-Epoxy-1,2,3,4,5,12-hexahydro-7-methoxy-1-(2-nitrophenyl)sulfonyl-2-naphthacenyl] Methyl Ketone (**16**). A mixture of **15** (490 mg, 1.21 mmol) and MVK (1 ml, 12.1 mmol) in anh. toluene (6 ml) was heated to 110° for 18 h. After evaporation, the residue was purified on column chromatography (SiO_2 , AcOEt/petroleum ether 1:1). The first fraction contained 55 mg (9%) of the minor adduct **17** and the second fraction the major adduct **16** which was recrystallized from CH_2Cl_2 / Et_2O 4:1 yielding 453 mg (79%) of yellow crystals, m.p. 189–191°. UV (CH_3CN): 210 (38500), 251 (40000), 284 (14000), 316 (3300), 331 (3400). IR (KBr): 3100, 3080, 3020, 2970, 2940, 2900, 2850, 1705, 1615, 1600, 1590, 1570, 1510, 1470, 1365, 1350, 1300, 1265, 1250, 1230, 1190, 1160, 1150, 1125, 1085, 1070, 1050, 930, 850, 835, 800, 780, 760, 740, 715, 690, 650, 640. $^1\text{H-NMR}$ (CD_2Cl_2): 8.15–7.28 (*m*, 6H, Ar-S, H–C(9), H–C(10)); 8.0 (*s*, H–C(6)); 7.33 (*s*, H–C(11)); 6.9 (*dd*, $^3J(\text{H–C}(8), \text{H–C}(9)) = 7.5$, $^4J(\text{H–C}(8), \text{H–C}(10)) = 1.0$, H–C(8)); 5.56 (br. s, $^5J(\text{H–C}(12), \text{H}_\beta\text{–C}(4)) = 1.0$,

H–C(12)); 5.53 (br. s, $^5J(\text{H–C}(5), \text{H}_\beta\text{–C}(1)) = 1.0$, H–C(5)); 3.92 (dddd, $^3J(\text{H}_\beta\text{–C}(1), \text{H–C}(2)) = 6.2$, $^5J(\text{H}_\beta\text{–C}(1), \text{H}_\alpha\text{–C}(4)) = 2.8$, $^5J(\text{H}_\beta\text{–C}(1), \text{H}_\beta\text{–C}(4)) = 2.2$, $^5J(\text{H}_\beta\text{–C}(1), \text{H–C}(5)) = 1.0$, $\text{H}_\beta\text{–C}(1)$); 4.0 (s, CH_3O); 2.71 (ddd, $^3J(\text{H–C}(2), \text{H}_\beta\text{–C}(3)) = 7.8$, $^3J(\text{H–C}(2), \text{H}_\beta\text{–C}(1)) = 6.2$, $^3J(\text{H–C}(2), \text{H}_\alpha\text{–C}(3)) = 4.0$, H–C(2)); 2.48 (m, $^2J = 18.5$, $^5J(\text{H}_\beta\text{–C}(4), \text{H}_\beta\text{–C}(1)) = 2.2$, $^5J(\text{H}_\beta\text{–C}(4), \text{H–C}(12)) = 1.0$, $\text{H}_\beta\text{–C}(4)$); 2.17 (s, CH_3CO); 2.16–2.06 (m, $^2J = 18.5$, $^5J(\text{H}_\alpha\text{–C}(4), \text{H}_\beta\text{–C}(1)) = 2.8$, $\text{H}_\alpha\text{–C}(4)$); 2.02–1.88 (m, $\text{H}_2\text{C}(3)$). MS (70 eV): 473 (1, M^+), 318 (10), 278 (19), 277 (100), 276 (19), 275 (17), 260 (10), 259 (14), 249 (14). Anal. calc. for $\text{C}_{27}\text{H}_{23}\text{NO}_3\text{S}$ (473.544): C 68.48, H 4.89; found: C 68.32, H 4.99.

(5-Acetoxy-3,4-dihydro-7-methoxy-2-naphthacenyl) Methyl Ketone (20). A soln. of **16** (156 mg, 0.33 mmol) in Ac_2O (4.5 ml) containing CF_3COOH (60 μl , 0.78 mmol) was heated to 50° under N_2 for 1 h. Anh. pyridine (1 ml) was added and the mixture allowed to cool to 20° . After stirring at 20° for 20 min, CH_2Cl_2 (10 ml) was added and the soln. extracted successively with 5% aq. HCl soln. (10 ml, 3 times) and sat. aq. K_2CO_3 soln. (10 ml, 3 times). After drying (MgSO_4), the solvent was evaporated and the residue taken up with CH_2Cl_2 (6 ml). A 0.5M aq. soln. of NaHCO_3 (6 ml) and MCPBA (90 mg, 0.47 mmol) was added and the mixture stirred at 20° for 15 min. The org. layer was separated and washed with sat. aq. K_2CO_3 soln. (10 ml, 3 times). After drying (MgSO_4), the soln. was filtered through silica gel (5 g) and the solv. evaporated. The residue contained a 3:1 to 4:1 mixture **18/19**. The major isomer **18** was obtained pure by recrystallization from $\text{CH}_2\text{Cl}_2/\text{Et}_2\text{O}$ 5:1 yielding 77 mg (65%) of yellow crystals, m.p. $199\text{--}201^\circ$ (dec). UV(CH_3CN): 239 (19000), 260 (12000), 306 (30000), 319 (31000), 361 (5200), 379 (5100). IR (KBr): 3065, 3020, 2960, 2910, 2850, 1760, 1655, 1635, 1615, 1565, 1460, 1440, 1385, 1370, 1310, 1275, 1255, 1225, 1205, 1170, 1155, 1110, 1065, 980, 910, 890, 790. $^1\text{H-NMR}$ (CD_2Cl_2): 8.69 (s, H–C(6)); 8.41 (s, H–C(11)); 7.86 (s, H–C(12)); 7.65 (s, H–C(1)); 7.59 (d, $^3J(\text{H–C}(10), \text{H–C}(9)) = 7.5$, H–C(10)); 7.42 (dd, $^3J = 7.5$, $^3J(\text{H–C}(9), \text{H–C}(8)) = 7.4$, H–C(9)); 6.82 (d, $^3J = 7.4$, H–C(8)); 4.09 (s, CH_3O); 2.99–2.59 (m, $\text{H}_2\text{C}(3), \text{H}_2\text{C}(4)$); 2.59 (s, OCOCH_3); 2.49 (s, 3H, COCH_3). Irradiation at 7.65 (H–C(1)) led to NOE's at 7.86 (H–C(12)) and 2.49 (COCH_3); 7.86→7.65 and 8.41; 8.41→7.86 and 7.59; 7.52→8.41 and 7.42; 4.09(CH_3O)→6.82(H–C(8)) and 8.69(H–C(6)); 2.59(OCOCH_3)→8.69(H–C(6)); 2.49(COCH_3)→7.65 ppm. MS (70 eV): 361 (7), 360 (32, M^+), 318 (100), 275 (24), 202 (9). MS (HR): 360.1360 ($\text{C}_{23}\text{H}_{20}\text{O}_4$, calc. 360.1361).

6-Acetoxy-9-acetyl-7,8-dihydro-4-methoxy-5,12-naphthacenequinone¹⁰ (22). A soln. of **20** (70 mg, 0.18 mmol) in acetone (8 ml) and 4N Jones' reagent (0.9 ml, 1.22 mmol of CrO_3 ; from 26.72 g of CrO_3 , 23 ml of conc. H_2SO_4 , diluted to 200 ml with H_2O) were added dropwise and simultaneously to stirred acetone (8 ml) maintained at 0° . After stirring at 0° for 1 h, *i*-PrOH (1 ml) was added and the mixture stirred for 10 min, then CH_2Cl_2 (20 ml) was added. The soln. was washed with 10% aq. NaHCO_3 soln. (10 ml, 2 times) and then with sat. aq. NaCl soln. (10 ml, 3 times). The aq. phase was extracted with CH_2Cl_2 (10 ml, 2 times). The org. extracts were united, dried (MgSO_4), and evaporated to dryness. The crude **22** was recrystallized from $\text{CH}_2\text{Cl}_2/\text{acetone}/\text{MeOH}$ 1:4:1 yielding 42 mg (64%) of yellow crystals, m.p. $238\text{--}240^\circ$ (dec). UV (CH_3CN): 233 (27000), 287 (53000), 297 (53000), 282 (12000). IR (KBr): 3100, 2950, 2850, 1750, 1670, 1590, 1475, 1455, 1440, 1390, 1370, 1355, 1335, 1320, 1300, 1280, 1265, 1250, 1235, 1210, 1195, 1180, 1120, 1105, 1050, 1020, 1010, 965, 925, 885, 795, 750. $^1\text{H-NMR}$ (CD_2Cl_2): 8.06 (s, H–C(11)); 7.90 (dd, $^3J(\text{H–C}(1), \text{H–C}(2)) = 7.7$, $^4J(\text{H–C}(1), \text{H–C}(3)) = 1$, H–C(1)); 7.7 (dd, $^3J = 8.5$, 7.7, H–C(2)); 7.48 (br. s, H–C(10)); 7.34 (dd, $^3J(\text{H–C}(2), \text{H–C}(3)) = 8.5$, $^4J(\text{H–C}(3), \text{H–C}(1)) = 1$, H–C(3)); 4.03 (s, CH_3O); 2.98–2.60 (m, $\text{H}_2\text{C}(7), \text{H}_2\text{C}(8)$); 2.55 (s, OCOCH_3); 2.51 (s, COCH_3). MS (70 eV): 390 (9, M^+), 349 (19), 348 (100), 333 (19), 332 (12), 330 (16), 315 (10), 305 (49), 291 (13), 290 (28), 289 (14), 288 (13), 287 (26). MS (HR): 390.1106 ($\text{C}_{23}\text{H}_{18}\text{O}_6$, calc. 390.1103).

(\pm)-6-Acetoxy-9-acetyl-9,10-epoxy-7,8,9,10-tetrahydro-4-methoxy-5,12-naphthacenequinone¹⁰ (23). A mixture of **22** (96 mg, 0.2 mmol) and MCPBA (471 mg, 2.73 mmol) in CH_2Cl_2 (5 ml) was stirred at 20° for 15 h. The precipitate was filtered off and the soln. washed with 10% aq. K_2CO_3 soln. (10 ml, 5 times). After drying (MgSO_4) and evaporation, the crude **23** was purified on silica gel ($\text{AcOEt}/\text{hexane}$ 4:1) yielding 42 mg (42%) of yellow-orange crystals, m.p. $235\text{--}6^\circ$ (dec.; $\text{CH}_2\text{Cl}_2/\text{Et}_2\text{O}$ 2:1). UV (CH_3CN): 219 (33000), 260 (35000), 281 (sh, 6700), 381 (6400). IR (KBr): 2980, 2940, 2880, 1775, 1715, 1680, 1590, 1440, 1380, 1370, 1350, 1330, 1220, 1200, 1180, 1110, 1070, 1040, 840. $^1\text{H-NMR}$ (CDCl_3): 8.25 (s, H–C(11)); 7.91 (dd, $^3J(\text{H–C}(1), \text{H–C}(2)) = 7.5$, $^4J(\text{H–C}(1), \text{H–C}(3)) = 1$, H–C(1)); 7.71 (dd, $^3J = 7.5$, 7.2, H–C(2)); 7.34 (dd, $^3J(\text{H–C}(3), \text{H–C}(2)) = 7.5$, $^4J(\text{H–C}(3), \text{H–C}(1)) = 1$, H–C(3)); 4.18 (s, H–C(10)); 4.02 (s, CH_3O); 2.75–2.65 (m, 2H); 2.47 (s, OCOCH_3); 2.34–2.22 (m, 2H); 2.17 (s, COCH_3). MS (70 eV): 406 (11, M^+), 380 (19), 364 (100), 363 (10), 362 (11), 349 (15), 348 (56), 347 (24), 346 (57), 338 (68), 328 (30), 322 (87), 321 (66), 320 (27), 304 (75). MS (HR): 406.1050 ($\text{C}_{23}\text{H}_{18}\text{O}_{17}$, calc. 406.1052).

¹⁰) IUPAC names: 11-acetoxy-8-acetyl-9,10-dihydro-1-methoxy-5,12-naphthacenequinone (**22**); 11-acetoxy-8-acetyl-7,8-epoxy-7,8,9,10-tetrahydro-1-methoxy-5,12-naphthacenequinone (**23**); 8-acetyl-7,8,9,10-tetrahydro-8,11-dihydroxy-1-methoxy-5,12-naphthacenequinone ((\pm) -**24**).

(\pm)-7,11-Dideoxydaunomycinone¹⁰) ((\pm)-**24**). Air was bubbled through a soln. of NaOH (38 mg, 0.95 mmol) and Na₂S₂O₄ (81 mg, 0.46 mmol) in H₂O (1.3 ml) for 10 min. The epoxide **23** (9 mg, 0.022 mmol) was added and the mixture stirred at 20° for 1 h. Air was bubbled through the mixture for 10 min. The mixture was acidified with 5% aq. HCl, and air was bubbled for another 10 min. The mixture was extracted with CH₂Cl₂ (5 ml, 3 times) and the extract washed with H₂O (5 ml, 3 times). After evaporation, the crude **24** was purified by TLC (SiO₂, AcOEt/hexane 4:1) yielding 3.1 mg (38%) of yellow-orange crystals, m.p. 209–211° ([11] [12c]: 209–211°; [10]: 210–211°). ¹H-NMR (CDCl₃): 13.43 (s, HO–C(6)); 7.98 (d, ³J = 8.3, H–C(1)); 7.76 (t, ³J = 8.3, H–C(2)); 7.56 (s, H–C(11)); 7.36 (d, ³J = 8.3, H–C(3)); 4.10 (s, CH₃O); 3.73 (s, HO–C(9)); 3.32 (d, ²J = 17, H–C(10)); 3.15 (dd, 1H, ²J = 18, ⁵J = 2) and 2.95 (m, 1H, H₂C(7)); 2.82 (dd, ²J = 17, ⁵J = 2 Hz, H–C(10)); 2.38 (s, COCH₃), 2.02 (m, H₂C(8)). MS (70 eV): 366 (14, M⁺), 348 (18), 323 (100), 305 (19), 293 (12), 290 (10).

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